### ARES NO<sub>x</sub> Sensor Development

#### **Los Alamos National Laboratory**

Eric L. Brosha, R. Mukundan, Roger Lujan and David Brown, Research Staff

Fernando H. Garzon, Sensor Project Leader

Materials Science and Technology Division
Electrochemical Materials and Devices Group, MST-11



# Project Description, Goals, and Objectives

- Develop new, lost cost, high-temperature solid-state sensors to detect NO<sub>x</sub> in the exhaust gas stream of a reciprocating engine.
  - Ceramic sensors that operate on electrochemical principles
- Sensors to control regenerative NO<sub>x</sub> catalyst systems.
- Sensors must be inexpensive, robust, and be able to detect NO<sub>x</sub> in regimes of PO<sub>2</sub>, T, [NOx] characteristic of current recip engine technology
- Lifetime, stability, sensitivity, response time, sulfur tolerance, device reproducibility, cost, etc.



#### Project Team - LANL

- LANL team has 10 years of experience in the development of gas sensor technology
  - Oxygen
  - Sulfur dioxide
  - Hydrocarbons
  - Carbon Monoxide
- LANL CRADA with USCAR to develop high temperature electrochemical sensors for OBD-II applications.
- FY 1997 2001
  - Sensors required to measure real-time NMHC concentrations post catalytic converter with minimum CO cross interference as possible replacement of dual O<sub>2</sub> sensor approach currently used.
  - Partners: Ford, GM, and Chrysler
- "Mixed Potential Hydrocarbon Sensor with Low Sensitivity to Methane and CO" submitted for U.S. Patent, Spring 2000.



#### Project Team - con't

- Previous LANL sensor work examples:
  - LDRD sensor development work
    - "Controlled Interface Gas Sensors", U.S. Patent applied for 2002.
  - Sulfur resistant O₂ combustion control sensors
    - Winner 1999 R&D 100 award in collaboration with Rosemount Analytical Co.
    - U.S. Patent no. 6,277,256
  - Amperometric response, lean burn O<sub>2</sub> sensor work in collaboration with Delphi Automotive
    - U.S. Patent no.s 5,695,624 & 5,543,025
- <u>University Collaboration</u>: ARES NO<sub>x</sub> sensor development work will be performed in collaboration with Dr. Eric Wachsman and his sensor team at the University of Florida



#### Impact on Goals of ARES Program

- Advanced reciprocating engines require new sensor technologies to reduce emissions and improve over-all efficiency.
  - NO<sub>x</sub> emissions monitoring and control
  - Engine control
  - Control of NO<sub>x</sub> regeneration catalysts and systems
- E.g. NO<sub>x</sub> emissions cannot exceed 50 150 ppm (0.1 g/hp-hr).
- A NO<sub>x</sub> sensor is needed to monitor and control emissions systems.
- Currently there is no suitable sensing technology available that is low-cost sulfur tolerant, long-lifetime, etc.



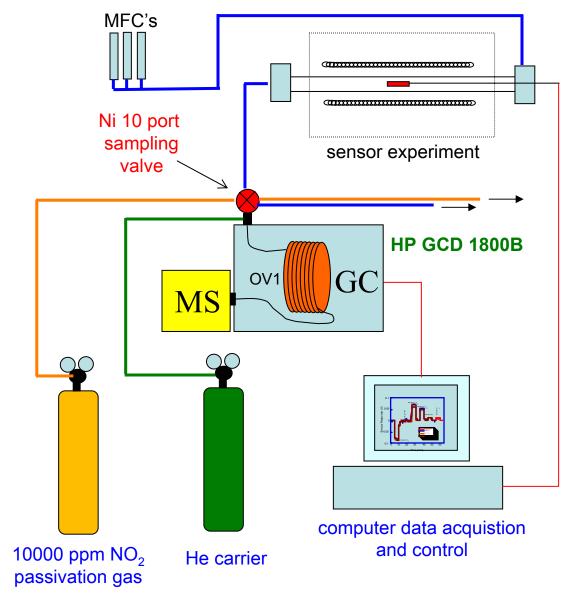
#### Milestones completed and planned

- Assemble analytical NOx capability at LANL with resources that are available.
   Completed
- Fabricate devices for initial study to demonstrate proof-of-concept. Completed
- Test and calibrate analytical [NO<sub>x</sub>] system and build new test stations. Underway
- Study factors that affect sensitivity, lifetime, response time.
- Study electrode materials NO<sub>x</sub> catalysis and electrocatalysis.
- Study electro-catalysis and NO<sub>x</sub> mixed potential electrochemistry.
- Improve NO<sub>x</sub> sensor.



#### Development of NO<sub>x</sub> Analysis Capability

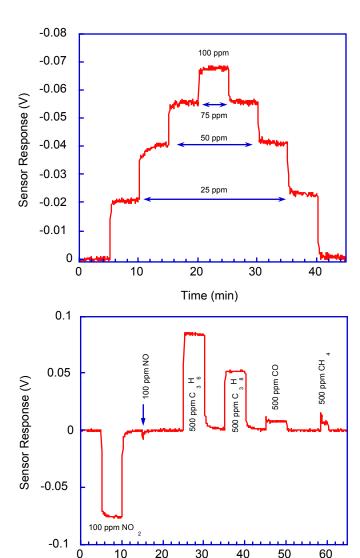
- Program code received February 2002.
- First task was to build an inhouse NO<sub>x</sub> measurement capability.
  - [NO<sub>x</sub>] changes over time
  - Complex equilibrium in presence of O<sub>2</sub>
- Capital equipment money to purchase commercial analytical NO<sub>x</sub> analysis instrumentation was not approved.
- We have obtained and made modifications to a salvaged GC/MS system that may allow us to characterize [NO<sub>x</sub>] and study catalysis.





#### **Preliminary Sensor Results**

- Different metal oxide electrode selected based on observations of higher NO<sub>x</sub> catalysis.
- Initial testing performed hot 650°C and in 1%O<sub>2</sub>
- Mixed potential response to NO<sub>2</sub> with no sensitivity to NO.
- Interference to CO and HC's
  - almost 5:1 sensitivity to NO<sub>2</sub>

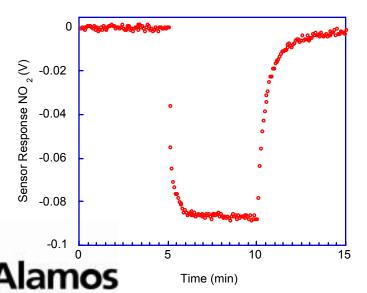


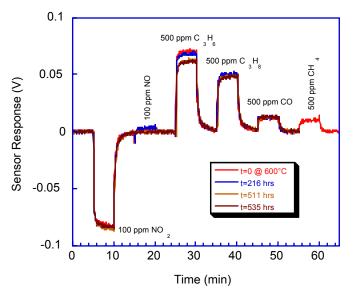
Time (min)

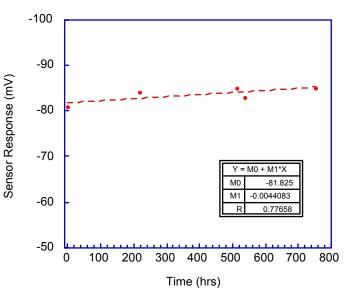


## Preliminary Sensor Results - level stability and response time

- Two mixed potential sensors are currently being studied for:
  - Stability 800 hrs
  - cross sensitivity C<sub>3</sub>H<sub>6</sub>, C<sub>3</sub>H<sub>8</sub>, CO, CH<sub>4</sub>
  - Reproducibility two devices
  - response time has been measured but test system has not been optimized to permit quantification.
    - sensors are faster than the experimental set-up.
- Data below: 600°C, 1%O<sub>2</sub>







#### Key Technical Barriers and Project Risks

- Key technical barriers are:
  - Designing sensors with adequate sensitivity and selectivity
  - Elimination of cross interferences
  - Response stability over sensor lifetime
    - Electrode ageing phenomena



#### Summary

- LANL NO<sub>x</sub> sensor work commenced in February 2002.
- Assembly of gas chromatography mass spectroscopy system for analytical NO<sub>x</sub> measurement complete. System is being tested and calibrated to see if method is viable to reliably measure concentrations of oxides of nitrogen.
- Fabricated several prototype mixed potential-based NO<sub>x</sub> sensors for characterization.
- Demonstrated  $NO_2$  response at  $1\%O_2$  and in air at  $[NO_2] = 25-100$  ppm
  - briefly examined level stability and potential for device fabrication with high device to device reproducibility - devices are excellent in both regards
- Expanding work to study factors which affect response time, NO<sub>x</sub> sensitivity, cross sensitivity, temperature coefficients, device reproducibility, etc.

